

Enhanced thermoelectric performance of CuGaTe₂ based composites incorporated with graphite nanosheets

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Abstract

CuGaTe₂ based composites incorporated with graphite nanosheets(GNs) CuGaTe₂/x G (G=GNs, 0≤x≤3.04 vol.%) were prepared, and the thermoelectric properties of the composites were studied from 300 to 875 K. The results show that the incorporation of GNs into the CuGaTe₂ matrix can enhance the Seebeck coefficient and power factor over the whole temperature range investigated due to energy filtering effects, and the reduction of thermal conductivity below 750K owing to interface scattering. Although the resistivity increases, energy filtering significantly raises the Seebeck component, and the overall effect on power factor is positive. The sample with 2.28 vol.% GNs had the largest ZT value, reaching 0.93 at 873K, which is a ~21% improvement on pure CuGaTe₂.

Today, thermoelectrics are attracting ever increasing interest from researchers due to their advantages in efficient and environmentally friendly energy conversion [1-4]. The efficiency of thermoelectric(TE) device is determined by the TE figure of merit, defined as $ZT = S^2\sigma T/\kappa$, where T, S, σ , and κ are the absolute temperature, Seebeck coefficient, electric conductivity, and thermal conductivity, respectively. A good TE material should possess a high power factor ($PF=S^2\sigma$) and low thermal conductivity.

The lead free material CuGaTe₂ has been widely investigated as a promising candidate for thin film solar cells, photovoltaic devices, and so on [5-11]. A p-type semiconductor with band gap of $E_g=1.2\text{eV}$, CuGaTe₂ has chalcopyrite structure of the space group I-42d. Until recently, it was reported as a promising thermoelectric (TE) candidate material. Theerayuth Plirdpring et al [12] found that pure CuGaTe₂ had high ZT value at high temperatures. To enhance the TE performance of CuGaTe₂, we introduced graphite nanosheets (GNs) into the matrix as a secondary phase. The aim

of this approach is to obtain a smaller thermal conductivity κ due to effective interface scattering ^[13-15], and to tailor the transport properties of CuGaTe₂ via enhancement of the carrier mobility through the injection of ultrahigh mobility electrons from the GNs^[16]. Experimental details are provided in the supplemental materials and in the Experimental section^[17].

Figure 1(a) shows the Raman spectra taken from GNs and the composites CuGaTe₂/x G (0≤x≤3.04 vol.%), three peaks marked with “D”, “G” and “2D” can be clearly identified, which are assigned to the D band (defects or edge areas) and G band (the vibration of sp² hybridized carbon) excitations^[18], which suggest the presence of GNs. The “2D” mode appears to be flat and weak as compared to the “G” mode, indicating that the materials may be multi layered graphene ^[19, 20]. Figure S1^[17] shows XRD patterns of the pure CuGaTe₂ and the composite samples, all the diffraction peaks can be indexed to the chalcopyrite structural phase of CuGaTe₂ (PDF#65-0244), indicating that all the specimens have the same crystallographic structure as pure CuGaTe₂ (see the insert in Figure S1^[17]). It can be seen from the HTEM image (Figure S2^[17] in the supplemental material), the thickness of the GN is about 2.8 nm, which indicates that the GNs are about 10 layers^[21]. Thus, the materials should be considered as graphite nanosheets not graphene according to the reference^[22]. Comparing the Raman spectrum of the GNs with the Raman spectrum of the composites, we can find that the relative intensity of “D”, “G” and “2D” peaks are similar, which indicates that the GNs are not damaged or significantly modified by their incorporation into CuGaTe₂ matrix. A Raman map of one of the composites, with GNs content x=3.04 vol.%, is shown in Figure 1(b), from which we can see the GNs are homogenous dispersed in the bulk material. The microstructures of GNs, CuGaTe₂ and CuGaTe₂/G composites were characterized by TEM and SEM, as shown in Figure 1(c)-(h). Figure 1(c) and (h) shows that the GNs exhibit transparency, indicating that they are quite thin. Figure 1(c) and Figure S2 indicate that the graphite sheets are about 2.8 nm thick and several to tens of microns in length and width. Figure 1(d) shows the ordered lattice of the CuGaTe₂ and the boundary between it and the GNs. From this image one can clearly see that the thin GNs are connected to the grain of CuGaTe₂ matrix, forming CuGaTe₂/GNs bulk composites. Figure 1(e) shows that the distribution of grain sizes is between tens of nanometers and several microns, which gives rise to more interface scattering, thus causing diminished thermal conductivity.

Figure 1(f)-(h) clearly indicate that the CuGaTe_2 crystals are aggregated with the GNs, as denoted by the white arrows, forming $\text{CuGaTe}_2/x\text{G}$ composites, in agreement with the result obtained from TEM analysis. Figure 1(h) shows that the CuGaTe_2 grains are covered with the transparent GNs. Generally, the CuGaTe_2 grain size becomes larger after the hot-pressing. However, SEM images of powdered and bulk (as shown in Figure 1(e)-(h)), the grain size for $\text{CuGaTe}_2/x\text{G}$ composites has no obviously changes after the hot-pressing process, which may lie in the fact that the addition of GNs limit the growth of matrix grains during the consolidation or sintering process. It can be seen from the SEM images of the bulk composites that grain size within the matrix is distributed over a wide range, from tens nanometers to several microns.

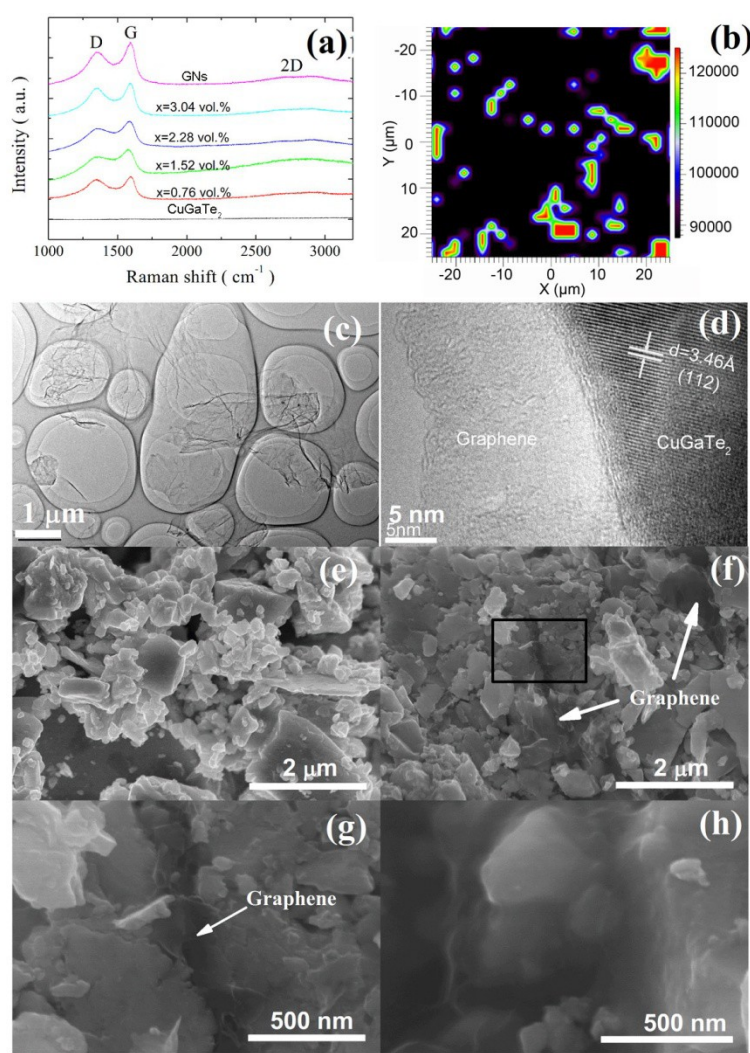


Figure 1 (a) Raman spectra of GNs and $\text{CuGaTe}_2/x\text{G}$ ($0 \leq x \leq 3.04 \text{ vol.}\%$). (b) Raman map at 1592 cm^{-1} for the composite $\text{CuGaTe}_2/ 3.04 \text{ vol.}\%\text{G}$. (c) TEM image of GNs. (d) HRTEM image of $\text{CuGaTe}_2/ 3.04 \text{ vol.}\%\text{G}$ composite. (e) SEM image of pure

CuGaTe₂ powder. (f) SEM image of bulk specimen CuGaTe₂/ 3.04 vol.%G composite after hot-pressing. (g) Enlarged view of the area within the rectangle in (f). (h) CuGaTe₂ particles covered with GNs in the bulk specimen of CuGaTe₂/ 3.04 vol.%G composite.

The thermoelectric properties of GNs dispersed CuGaTe₂ composites are summarized in Figures 2-4. As shown in Figure 2(a), ρ for pure CuGaTe₂ increases slightly with increasing temperature until it reaches a maximum at $T=492\text{K}$, and decreases beyond this temperature ($d\rho/dT < 0$), indicating that CuGaTe₂ is a partial degenerate semiconductor^[23]. In contrast, the resistivity ρ for the CuGaTe₂/xG composites decreases steeply with increasing temperature, demonstrating its non-degenerate semiconducting behavior. In addition, the ρ values for the all composites CuGaTe₂/xG ($x > 0$) are larger than for pure CuGaTe₂ over the whole temperature range investigated. For example, the ρ value increases from $7.9 \times 10^{-5} \Omega \cdot \text{m}$ to 2.9×10^{-4} , 2.9×10^{-4} , 3.0×10^{-4} and $32 \times 10^{-4} \Omega \cdot \text{m}$ as the at room temperature (RT) GNs content x increases from 0 to 0.76%, 1.52%, 2.28% and 3.04%, respectively. A threefold increase in ρ is observed as x is increased from 0 to 3.04% at RT. The transition from a degenerate to a non-degenerate state, and the increase in ρ with the addition of GNs can be attributed to the remarkable decrease in hole concentration caused by introducing the second phase. This is verified by measurements of the Hall carrier concentration p , as shown in Figure 3(a). The holes concentration p decreased from the order of 10^{19} to the order of 10^{18} cm^{-3} after GNs addition. The decrease in the hole concentration can lead to the Fermi level E_F shifting away from valence band, resulting in a non-degenerated semiconductor. Electrons are introduced from the GNs into the CuGaTe₂ matrix, which causes electron-hole recombination, and decreasing the hole concentration. The concentration of holes p at room temperature in pure CuGaTe₂ is $1.49 \times 10^{19} \text{ cm}^{-3}$, which is different from reported values ($p=1.1 \times 10^{18} \text{ cm}^{-3}$ ^[12], $p=4.6 \times 10^{18} \text{ cm}^{-3}$ ^[24] and $p=2.05 \times 10^{19} \text{ cm}^{-3}$ ^[25]). These differences may due to different sample compositions, such as the different Te content, which would lead to a distinct difference in the hole concentration p and thermo-power S values^[12]. In fact, the value of the hole concentration at room temperature decreases rapidly from $1.49 \times 10^{19} \text{ cm}^{-3}$ to $3.86 \times 10^{18} \text{ cm}^{-3}$, $3.49 \times 10^{18} \text{ cm}^{-3}$, $2.57 \times 10^{18} \text{ cm}^{-3}$ as the GNs content x increases from 0 to 0.76, 1.52 and 2.28 vol.%. However, the hole concentration p

increases slightly to $3.08 \times 10^{18} \text{ cm}^{-3}$ when GNs content x is further increased to 3.04 vol.%. It is worth noting an interesting phenomenon here is that there is no reduction or even increase in the hole mobility μ as the GNs are incorporated. This is despite GNs being known for its high mobility, which is in contrast to the low mobility of p -type CuGaTe₂. High-mobility carriers from the GNs are injected into the low-mobility CuGaTe₂ matrix, hence enhancing the overall carrier mobility^[16]. Another noteworthy feature is demonstrated in Figure 3(a), while the carrier concentration reaches a minimum as $x=2.28 \text{ vol.}\%$, the mobility attains its highest value, and this is also where the ZT reaches its maximum value.

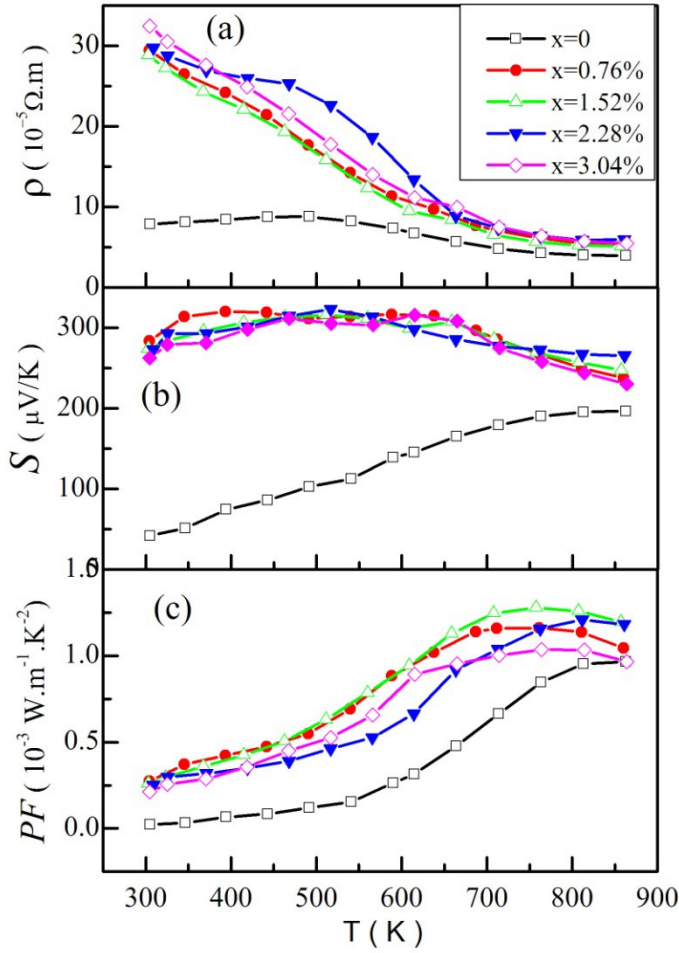


Figure 2 Temperature dependence of (a) the electrical resistivity ρ , (b) thermopower S and (c) the power factor PF for CuGaTe₂/ x G composites ($0 \leq x \leq 3.04 \text{ vol.}\%$).

As shown in Figure 2(b), the positive values of S over the whole temperature range indicate that the major charge carriers are holes. Moreover, one can see that S for the pure CuGaTe₂ sample increases linearly with increasing temperature within the range of 300K to ~750K, showing its degenerate behavior. As T increases further, S increases more slowly. However, S values for all the composite samples are larger over the wide temperature range of 300-600K, beyond which S decreases with increasing the temperature. One notices that there are no large differences in S for the samples with different concentrations of incorporated GNs, which is analogous to the behavior of the resistivity. As shown in Figure 2(c), the power factor (PF) for the composite samples are larger than of the pure CuGaTe₂ over the whole temperature range investigated. The maximum PF achieved is $1.28 \times 10^{-3} \text{ Wm}^{-1}\text{K}^{-2}$ at 758K for the sample with $x=1.52 \text{ vol.}\%$, which is ~54% larger than the value for pure CuGaTe₂ at the same temperature due to the enhanced S .

S is inversely proportional to carrier concentration p , as expressed by the Mott equation:

(1)

where σ is the electrical conductivity, q is the carrier charge, k_B is the Boltzmann constant, and E_F is the Fermi energy. Assuming the energy bands have a parabolic dispersion relation, the relaxation time $\tau(E)$, which is usually in the form of a power, with energy E , i.e., $\tau(E)=\tau_0 E^{\lambda-1/2}$ ^[26]

then, Equation (1) can be expressed as:

(2)

From Eq. (2) it can be understood that S can be enhanced by increasing the scattering parameter λ . An enhanced λ means a stronger energy filtering effect. In order to verify that energy filtering effect is responsible for obviously enhancement of S , λ was calculated by using a single parabolic band model in which effective mass m^* and S can be expressed as:

(3)

(4)

with a Fermi integral of order i :

(5)

where h is Planck's constant and ξ_F is the reduced Fermi level $F_F/(k_B T)$.

For a degenerate semiconductor, such as pure CuGaTe₂, the relationship between the Seebeck coefficient S and the carrier concentration p obeys following equation [27]:

(6)

Here m^* is the effective mass of the carrier. $m^*=0.28 m_0$ (here m_0 is the free electron mass) can be obtained for CuGaTe₂ according equation (6) and S , p values from Figure 3. Using the values of Seebeck coefficient S , the hole concentration p at room temperature, and assuming effective mass m^* is constant in the different composite samples[28], we can obtain the scattering parameters λ according equations (2)-(5), as shown in Figure 3(b). It can be seen that all the λ values of the composite samples are larger than that of pure CuGaTe₂. For example, λ increases from zero to 1.34 with increasing x from 0 to 0.76 vol. %. This increase in λ results in increases of ~181, 164, 138 and 143 $\mu V/K$ in S at room temperature for the samples with $x=0.76$, 1.52, 2.28 and 3.04 vol.%, respectively, as shown in Figure 3(c) (where the solid line is the Pisarenko relationship[27] for CuGaTe₂ and shows the dependence of S on p as calculated according equation (6)). The tendency of S to increase in accordance with the scattering parameter λ determined from the GNs content. In our work, the thermoelectric performance of CuGaTe₂ was enhanced by energy filtering due to nanosized potential barriers (GNs). Analogous work has been performed by Neophytou et al[29] and Yang et al[30]. And ones reported the energy filtering effect in the low-dimensional materials, for example, nanocrystalline[31], superlattices[32, 33], quantum well[34] and one-dimensional nanowires[34, 35]. Neophytou et al[29] reported analysis and optimization of such barriers for improved energy filtering, and Yang et al reported synergetic scattering of electrons in semiconductors with metal nanoinclusions. In agreement with Neophytou et al[29], to maximize energy filtering, in the region prior to the barrier, the Fermi level needs to be pushed up so that carriers have higher velocity and that scattering by ionized dopant impurities will be weaker, which are achieved in our work. Thus, the enhanced thermopower can be ascribed to energy filtering. Like the work reported by Heremans et al [36], we assume that energy

bands have a parabolic dispersion relation, $\tau(E)$ usually has the power form with energy E , i.e., $\tau(E)=\tau_0 E^{\lambda-1/2}$ and the effective mass is a constant for the composites and pure matrix, moreover, the effective mass is determined by S - p dependence.

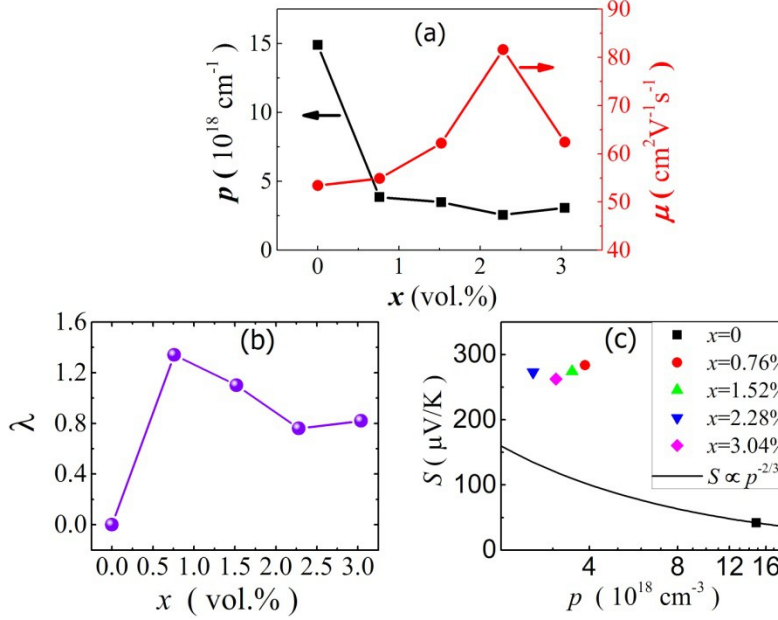


Figure 3 (a) The dependence of carrier concentration and carrier mobility on GNs content, at room temperature. (b) Variation of scattering parameter λ with GNs content x at room temperature. (c) variation of Seebeck coefficient with carrier concentration for $\text{CuGaTe}_2/x \text{ G}$ ($0 \leq x \leq 3.04 \text{ vol.}\%$) composites at room temperature. The solid line is the Pisarenko relation for pure CuGaTe_2 .

Figure 4(a) shows the temperature dependence of the thermal conductivity κ for the composites. The κ of the composite samples decreases after GNs incorporation over almost the whole investigated temperature range. For example, κ decreases from $4.3 \text{ Wm}^{-1}\text{K}^{-1}$ to 3.36, 3.34, 3.23 and $3.20 \text{ Wm}^{-1}\text{K}^{-1}$ at room temperature with increasing the GNs content x from zero to 0.76, 1.52, 2.28 and 3.04 vol.%, respectively. The total thermal conductivity κ can be expressed by the sum of the lattice component (κ_l) and the hole component (κ_h) as $\kappa=\kappa_l+\kappa_h$. The κ_h values can be estimated from Wiedemann-Franz's law, $\kappa_h=LT/\rho$ (here L is the Lorentz number, $L=1.5 \times 10^{-8} \text{ V}^2 \text{ K}^{-2}$ [12] for $\text{CuGaTe}_2/x \text{ G}$ ($0 \leq x \leq 3.04 \text{ vol.}\%$)). Consequently, κ_l can be obtained from κ and κ_h , as shown in Figure 4(b). It can be seen by comparing Figure 4(a) with Figure 4(b) that

the values and temperature trend of the lattice conductivity κ_L are similar to those of the total thermal conductivity κ . These results indicate that thermal conductivity of the samples comes mainly from the lattice conductivity, and thus the decrease of the thermal conductivity can be mainly attributed to the reduction in the lattice thermal conductivity caused by interface scattering due to incorporation of the GNs.

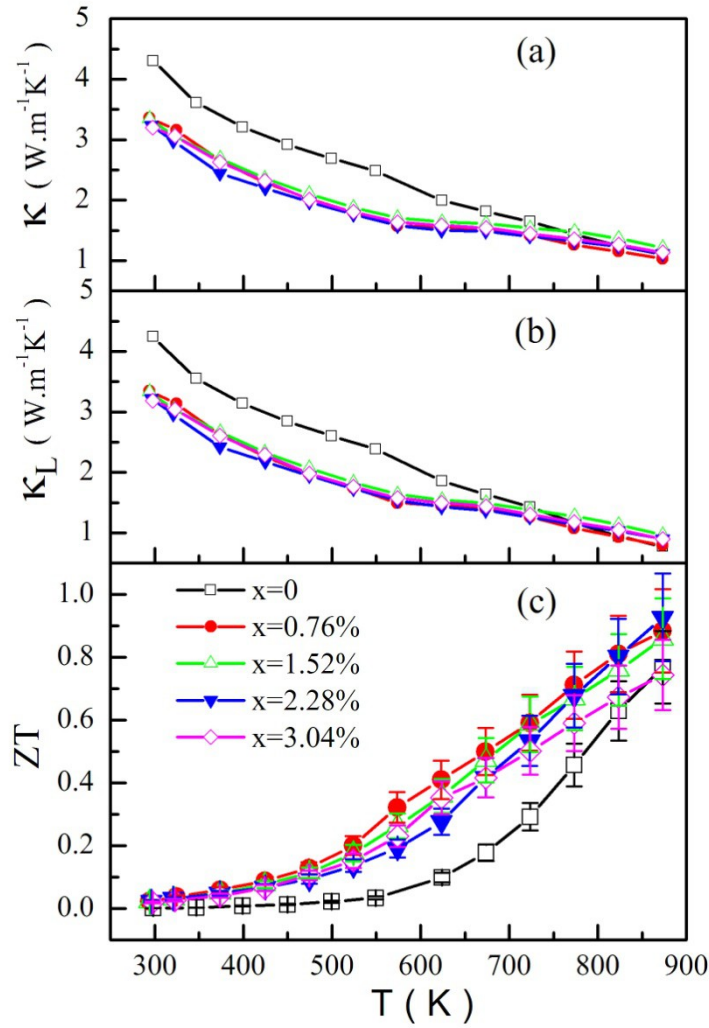


Figure 4 Plot of (a) total thermal conductivity κ , (b) lattice thermal conductivity κ_L and (c) ZT value versus temperature for CuGaTe_2/x G composites ($0 \leq x \leq 3.04 \text{ vol.}\%$)

The temperature dependence of the ZT values for CuGaTe_2/x G ($0 \leq x \leq 3.04 \text{ vol.}\%$) are shown in Figure 4(c). Due to both the increase in PF and the decrease in κ , the ZT of all the composite samples are enhanced as compared to pure CuGaTe_2 . One can see that the ZT values of all samples increase with increasing temperature over whole investigated temperature range. The ZT of the sample where $x=2.28\%$ has the

maximum value, reaching 0.93 at 873K, which is ~21% larger than for pure CuGaTe₂ at the same temperature (ZT=0.77 at 873K).

The thermoelectric properties of composites CuGaTe₂/xG (0 ≤ x ≤ 3.04 vol.%) have been studied for temperatures from 300 to 875 K. The results show that incorporation of GNs into the CuGaTe₂ matrix can concurrently yield a reduction of the thermal conductivity and an enhancement of the PF. The increased PF mainly comes from enhanced *S* due to an intensified energy filtering effect, and a moderate increase in carrier mobility owing to the injection of high mobility carriers from the GNs. The sample CuGaTe₂/2.28 vol.%G had the largest ZT value, reaching 0.93 at 873K, which is ~21% larger than for pure CuGaTe₂ at the same temperature. The results indicate that CuGaTe₂/2.28 vol.%G is a promising candidate as a thermoelectric material in the middle temperature range.

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References:

1. P. Puneet, R. Podila, S. Zhu, M. J. Skove, T. M. Tritt, J. He and A. M. Rao, *Adv Mater* **25**, 1033-1037 (2013).
2. B. C. Sales, *Science* **295**, 1248-1249 (2002).
3. W.-H. Chen, S.-R. Huang and Y.-L. Lin, *Appl Energ* **158**, 44-54 (2015).
4. A. Heghmanns and M. Beitelschmidt, *Appl Energ* **155**, 447-454 (2015).
5. M. León, R. Díaz, F. Rueda and M. Berghol, *Sol Energ Mat Sol C* **26**, 295-307 (1992).
6. D. Xia, S. Jiang, C. Liu, S. Fan and L. Chen, *Sol Energ Mat Sol C* **141**, 331-336 (2015).
7. Y. Zhang and Y. Xuan, *Sol Energ Mat Sol C* **144**, 68-77 (2016).
8. H. Park, S. J. Lee, S. Kim, H. W. Ryu, S. H. Lee, H. H. Choi, I. W. Cheong and J. H. Kim, *Polymer* **54**, 4155-4160 (2013).
9. B. Kuhn, W. Kaefer, K. Fess, K. Friemelt, C. Turner, M. Wendl and E. Bucher, *Phys. Status Solidi A-Appl. Res.* **162**, 661-671 (1997).
10. J. S. Im, J. Yun, J. G. Kim and Y. S. Lee, *Curr Org Chem* **17**, 1424-1433 (2013).
11. J. Q. Li, L. F. Li, S. H. Song, F. S. Liu and W. Q. Ao, *J Alloy Compd* **565**, 144-147 (2013).

12. T. Plirdpring, K. Kurosaki, A. Kosuga, T. Day, S. Firdosy, V. Ravi, G. J. Snyder, A. Harnwunggmoung, T. Sugahara, Y. Ohishi, H. Muta and S. Yamanaka, *Adv Mater* **24**, 3622-3626 (2012).
13. J. Y. Cho, Z. Ye, M. M. Tessema, R. A. Waldo, J. R. Salvador, J. Yang, W. Cai and H. Wang, *Acta Mater* **60**, 2104-2110 (2012).
14. T. Zou, X. Qin, Y. Zhang, X. Li, Z. Zeng, D. Li, J. Zhang, H. Xin, W. Xie and A. Weidenkaff, *Scientific Reports* **5**, 17803 (2015).
15. Y. Y. Li, X. Y. Qin, D. Li, X. Y. Li, Y. F. Liu, J. Zhang, C. J. Song and H. X. Xin, *Rsc Advances* **5**, 31399-31403 (2015).
16. Y. G. Y. W.J. Xie, S. Zhu, M. Zhou, S. Populoh, K. Gałazka, S.J. Poon, A. Weidenkaff, J. He, X.F. Tang, T.M. Tritt, *Acta Mater* **61**, Pages 2087-2094 (2013).
17. Y. C. Dou, X. Y. Qin, D. Li, Y. Y. Li, H. X. Xin, J. Zhang, Y. F. Liu, C. J. Song and L. Wang, *RSC Advances* **5**, 34251-34256 (2015).
18. Y. Lu, Y. Song and F. Wang, *Mater Chem Phys* **138**, 238-244 (2013).
19. K. J. Norris, M. Garrett, E. Coleman, G. S. Tompa, J. Zhang and N. P. Kobayashi, *J Cryst Growth* **406**, 41-47 (2014).
20. X. Li, W. Cai, J. An, S. Kim, J. Nah, D. Yang, R. Piner, A. Velamakanni, I. Jung, E. Tutuc, S. K. Banerjee, L. Colombo and R. S. Ruoff, *Science* **324**, 1312-1314 (2009).
21. E. Yoo, J. Kim, E. Hosono, H.-s. Zhou, T. Kudo and I. Honma, *Nano Lett* **8**, 2277-2282 (2008).
22. N. Jung, N. Kim, S. Jockusch, N. J. Turro, P. Kim and L. Brus, *Nano Lett* **9**, 4133-4137 (2009).
23. A. Kosuga, T. Plirdpring, R. Higashine, M. Matsuzawa, K. Kurosaki and S. Yamanaka, *Appl. Phys. Lett.* **100**, 042108 (2012).
24. J. Zhang, X. Qin, D. Li, H. Xin, C. Song, L. Li, X. Zhu, Z. Wang, G. Guo and L. Wang, *Journal of Materials Chemistry A* **2**, 2891-2895 (2014).
25. Y. P. Li, Q. S. Meng, Y. Deng, H. Zhou, Y. L. Gao, Y. Y. Li, J. F. Yang and J. L. Cui, *Appl Phys Lett* **100** (2012).
26. X. H. Yang, X. Y. Qin, J. Zhang, D. Li, H. X. Xin and M. Liu, *Journal of Alloys and Compounds* **558**, 203-211 (2013).
27. J. L. Cui, Y. P. Li, Z. L. Du, Q. S. Meng and H. Zhou, *J Mater Chem A* **1**, 677-683 (2013).
28. T. H. Zou, X. Y. Qin, D. Li, G. L. Sun, Y. C. Dou, Q. Q. Wang, B. J. Ren, J. Zhang, H. X. Xin and Y. Y. Li, *Applied Physics Letters* **104**, 013904 (2014).
29. N. Neophytou and H. Kosina, *J Appl Phys* **114**, 044315 (2013).
30. X. H. Yang and X. Y. Qin, *Applied Physics Letters* **97** (2010).
31. N. Neophytos, Z. Xanthippi, K. Hans, F. Stefano, L. Bruno and N. Dario, *Nanotechnology* **24**, 205402 (2013).
32. W. Kim, S. L. Singer, A. Majumdar, D. Vashaee, Z. Bian, A. Shakouri, G. Zeng, J. E. Bowers, J. M. Zide and A. C. Gossard, *Appl Phys Lett* **88**, 242107 (2006).
33. D. Vashaee and A. Shakouri, *Phys Rev Lett* **92**, 106103-106103 (2004).
34. J. E. Cornett and O. Rabin, *Appl Phys Lett* **100**, 242106 (2012).
35. R. Kim and M. S. Lundstrom, *J Appl Phys* **110**, 034511 (2011).
36. J. P. Heremans, C. M. Thrush and D. T. Morelli, *Phys Rev B* **70**, 115334 (2004).